

## SYNTHESIS OF 3-(5-ARYL-1,3,4-OXADIAZOL-2-YL)CHROMONES

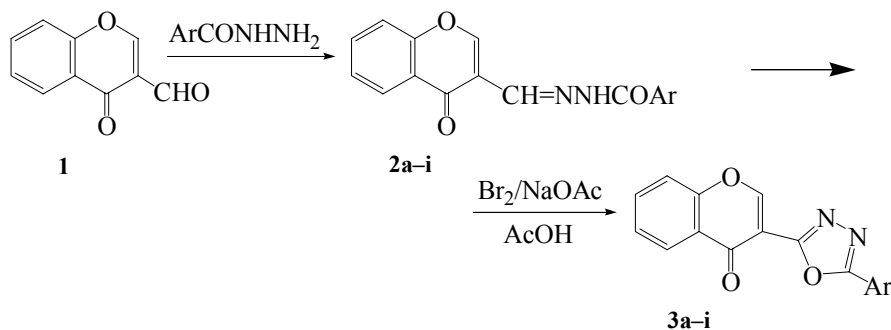
L. Cao<sup>1,2</sup> and W. Wang<sup>1</sup>

A general method was proposed for the synthesis of 3-(5-aryl-1,3,4-oxadiazol-2-yl)chromones starting from 3-formylchromone. This aldehyde yields acylhydrazones, from which the corresponding unstable nitrile imines are generated. These intermediates undergo intramolecular 1,3-dipolar cycloaddition.

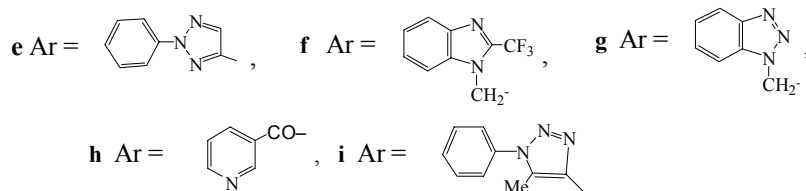
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3-Hetarylchromones have a broad range of biological activity and display high anti-allergy, anti-cholesterol, hypolipidemic, antimicrobial, fungicidal, and antitumor activity. Some of these compounds are also central nervous system stimulants [1]. Hence, considerable recent attention has been given to new compounds of this series.

Frasinyuk and Khilya [1] have surveyed the methods for synthesizing 3-hetarylchromones and found two major approaches: 1) construction of the chromone system from substituted  $\alpha$ -hetaryl-2-hydroxyacetophenones with participation of suitable reagents and 2) introduction of the heterocycle into a prepared chromone system. In the present work, we selected the second approach for the synthesis of previously unreported 3-(5-aryl-1,3,4-oxadiazol-2-yl)chromones starting from available 3-formylchromone (**1**) [2].



**2, 3 a** Ar = Ph, **b** Ar = *o*-ClC<sub>6</sub>H<sub>4</sub>, **c** Ar = *o*-(O<sub>2</sub>N)C<sub>6</sub>H<sub>4</sub>, **d** Ar = *p*-(O<sub>2</sub>N)C<sub>6</sub>H<sub>4</sub>,



<sup>1</sup> Chemistry Faculty, Sinsiang University, 830046 Urumchi, Chinese People's Republic; e-mail: clhxj@xju.edu.cn. <sup>2</sup> State Central Laboratory, Nankai University, 300070 Tientsin, Chinese People's Republic. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 8, pp. 1227-1230, August, 2003. Original article submitted October 21, 2002.

The reaction of **1** and aroylhydrazines gave the corresponding aroylhydrazones **2a-i**. Then, the action of bromine in the presence of sodium acetate on these hydrazones **2** gave intermediate nitrile imines [3], which undergo intramolecular cycloaddition to give 3-(5-aryl-1,3,4-oxadiazol-2-yl)chromones **3**.

The IR spectra of aroylhydrazones **2a-i** have characteristic bands at 3200-3300 (NH) and 1630-1650  $\text{cm}^{-1}$  (C=O). The  $^1\text{H}$  NMR spectra of aroylhydrazones **2a-i** show signals at 8.2-8.9 (pyrone 2-H) and 9.3-10.2 ppm (NH). The molecular ion peak in the mass spectra of these compounds has low intensity due to their instability. The appearance of peaks with mass number 187 and  $\text{ArCO}^+$  peaks indicates that the C(O)-N bond is readily cleaved and the major peak usually corresponds to the  $\text{ArCO}^+$  fragment except for **2g**, for which the  $\text{C}_6\text{H}_5^+$  peak with mass number 77 predominates. The pyrone ring is split by a reverse Diels-Alder reaction, leading to a fragment with  $m/z$  120. The stepwise loss of one CO group and formation of fragments with  $m/z$  92 and 64 is then noted.

The IR spectra of 3-(5-aryl-1,3,4-oxadiazol-2-yl)chromones **3a-i** lack the band at 3200-3300  $\text{cm}^{-1}$  found for hydrazones **2**. The  $^1\text{H}$  NMR spectra of **3a-i** lack signals at 9.3-10.2 ppm. Proton 2-H of the pyrone ring is seen as a narrow signal at 5.8-6.1 ppm since it is affected by the unshared electron pair of the heteroatom of the heterocyclic substituent at C-3 of the chromone.

Chromones **3a-a** have a stable conjugation system. Thus, the molecular ion is the major peak in their mass spectra. As a consequence, the oxadiazole ring is readily cleaved and a strong peak with  $m/z$  173 ( $[\text{M} - \text{ArCN}_2]^+$ ) is noted.

## EXPERIMENTAL

GF-254 Plates were used for thin-layer chromatography. The melting points were determined using an MP-S3 table manufactured in Japan. An MT-3 automatic analyzer was used for elemental analysis. The IR spectra were taken on a Bruker EQUINOX-55 FT-IR spectrometer for KBr pellets. The  $^1\text{H}$  NMR spectra were taken on a Bruker AX-200 spectrometer at 200 MHz in  $\text{CDCl}_3$  or  $\text{DMSO}-d_6$  with TMS as the internal standard. The mass spectra were taken on a HP 5988 automatic mass spectrometer.

**General Method for the Preparation of Aroylhydrazones 2a-i.** 3-Formylchromone **1** and aroylhydrazones obtained taking account of the work of Chen [4] were mixed in equivalent amounts and then dissolved in 95% aq. ethanol. A few drops of glacial acetic acid were added and the mixture was heated at reflux for 5-6 h. After cooling, the crystalline precipitate was filtered off and recrystallized from abs. ethanol to give aroylhydrazones **2a-i**.

**Hydrazone 2a** was obtained in 78% yield; mp 176-178°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3280 (-NH-), 3076 (Ar-H), 1679 (C=O), 1636 (C=N), 1070 (C-O-C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 10.21 (1H, br. s, NH); 8.87 (1H, s, 2-H); 7.24-7.98 (10H, m, CH=N, Ar-H). Mass spectrum,  $m/z$  292  $[\text{M}]^+$ . Found, %: C 70.31; H 4.19; N 9.68.  $\text{C}_{17}\text{H}_{12}\text{N}_2\text{O}_3$ . Calculated, %: C 69.86; H 4.14; N 9.58.

**Hydrazone 2b** was obtained in 80% yield; mp 193-195°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3230 (-NH-), 3059 (Ar-H), 1674 (C=O), 1633 (C=N), 1059 (C-O-C), 656 (C-Cl).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 10.15 (1H, br. s, NH); 8.77 (1H, s, 2-H); 7.23-7.92 (9H, m, CH=N, Ar-H). Mass spectrum,  $m/z$  328, 326  $[\text{M}]^+$ . Found, %: C 61.89; H 3.41; N 8.67.  $\text{C}_{17}\text{H}_{11}\text{ClN}_2\text{O}_3$ . Calculated, %: C 62.49; H 3.39; N 8.57.

**Hydrazone 2c** was obtained in 95% yield; mp 206-207°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3267 (-NH-), 3079 (Ar-H), 1666 (C=O), 1633 (C=N), 1520, 1341 ( $-\text{NO}_2$ ), 1071 (C-O-C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.74 (1H, br. s, NH); 8.70 (1H, s, 2-H); 8.42 (1H, d, CH=N); 7.23-7.92 (8H, m, Ar-H). Mass spectrum,  $m/z$  337  $[\text{M}]^+$ . Found, %: C 60.63; H 3.31; N 12.40.  $\text{C}_{17}\text{H}_{11}\text{N}_3\text{O}_5$ . Calculated, %: C 60.54; H 3.29; N 12.46.

**Hydrazone 2d** was obtained in 90% yield; mp 230-232°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3260 (-NH-), 3071 (Ar-H), 1670 (C=O), 1636 (C=N), 1498, 1319 ( $-\text{NO}_2$ ), 1518, 1338 (C-O-C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.72 (1H, br. s, NH); 8.76 (1H, s, 2-H); 8.47 (1H, d, CH=N); 7.25-7.98 (8H, m, Ar-H). Mass spectrum,  $m/z$  337  $[\text{M}]^+$ . Found, %: C 60.32; H 3.21; N 12.58.  $\text{C}_{17}\text{H}_{11}\text{N}_3\text{O}_5$ . Calculated, %: C 60.54; H 3.29; N 12.45.

**Hydrazone 2e** was obtained in 54% yield; mp 250-251.5°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3251 (–NH–), 3043 (Ar–H), 1636 (C=O), 1617 (C=N), 1097 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.83 (1H, br. s, NH); 8.35 (1H, s, 2-H); 8.20 (1H, s, triazole H); 7.25-8.03 (10H, m, Ar–H). Mass spectrum,  $m/z$  359  $[\text{M}]^+$ . Found, %: C 64.19; H 3.69; N 19.79.  $\text{C}_{19}\text{H}_{13}\text{N}_5\text{O}_3$ . Calculated, %: C 63.51; H 3.65; N 19.49.

**Hydrazone 2f** was obtained in 85% yield; mp 225-227°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3220 (–NH–), 2994 (Ar–H), 1698 (C=O), 1639 (C=N), 1119 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.32 (1H, br. s, NH); 8.58 (1H, s, 2-H); 8.31 (1H, d, CH=N); 7.24-7.92 (8H, m, Ar–H); 5.94 (2H, s,  $\text{CH}_2$ ). Mass spectrum,  $m/z$  414  $[\text{M}]^+$ . Found, %: C 57.95; H 3.20; N 13.63.  $\text{C}_{20}\text{H}_{13}\text{F}_3\text{N}_4\text{O}_3$ . Calculated, %: C 57.98; H 3.16; N 13.52.

**Hydrazone 2g** was obtained in 61% yield; mp 240-241°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3219 (–NH–), 2990 (Ar–H), 1698 (C=O), 1640 (C=N), 1109 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.63 (1H, br. s, NH); 8.52 (1H, s, 2-H); 8.30 (1H, d, CH=N); 7.23-7.86 (8H, m, Ar–H); 5.52 (2H, s,  $\text{CH}_2$ ). Mass spectrum,  $m/z$  347  $[\text{M}]^+$ . Found, %: C 62.20; H 3.80; N 20.21.  $\text{C}_{18}\text{H}_{13}\text{N}_5\text{O}_3$ . Calculated, %: C 62.25; H 3.77; N 20.16.

**Hydrazone 2h** was obtained in 90% yield; mp 214-216°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3262 (–NH–), 3093 (Ar–H), 1682 (C=O), 1648 (C=N), 1046 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.96 (1H, br. s, NH); 8.92, 8.81 (2H, Py–H); 8.2 (1H, s, 2-H); 7.33-7.74 (7H, m, Ar–H, Py–H). Mass spectrum,  $m/z$  293  $[\text{M}]^+$ . Found, %: C 66.04; H 3.83; N 14.53.  $\text{C}_{16}\text{H}_{11}\text{N}_3\text{O}_3$ . Calculated, %: C 65.53; H 3.78; N 14.33.

**Hydrazone 2i** was obtained in 92% yield; mp 223.5-225°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3240 (–NH–), 3069 (Ar–H), 1677 (C=O), 1641 (C=N), 1059 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 9.56 (1H, br. s, NH); 8.5 (1H, s, 2-H); 8.25 (1H, d, CH=N); 7.56-7.26 (9H, m, Ar–H); 2.69 (3H, s,  $\text{CH}_3$ ). Mass spectrum,  $m/z$  373  $[\text{M}]^+$ . Found, %: C 64.56; H 3.98; N 18.82.  $\text{C}_{20}\text{H}_{15}\text{N}_5\text{O}_3$ . Calculated, %: C 64.34; H 4.05; N 18.76.

**General Method for the Preparation of 3-(5-Aryl-1,3,4-oxadiazol-2-yl)chromones 3a-i.** A sample of bromine (2.2 mmol) in glacial acetic (5 ml) was added dropwise to a solution of aroylhydrazone **2a-i** (2 mmol) and fused sodium acetate (10 mmol) in glacial acetic acid (10 ml) and stirred vigorously for 2-3 h at room temperature. The reaction mixture was poured into chopped ice. The precipitate formed was recrystallized from absolute ethanol to give chromones **3a-i**.

**3-(5-Phenyl-1,3,4-oxadiazol-2-yl)chromone (3a)** was obtained in 62% yield; mp 224-226°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3044 (Ar–H), 1683 (Ar–H), 1683 (C=O), 1629 (C=N), 1029 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 7.24-7.99 (9H, m, Ar–H); 5.93 (1H, s, 2-H). Mass spectrum,  $m/z$  290  $[\text{M}]^+$ . Found, %: C 70.66; H 3.51; N 9.73.  $\text{C}_{17}\text{H}_{10}\text{N}_2\text{O}_3$ . Calculated, %: C 70.34; H 3.47; N 9.65.

**3-[5-(*o*-Chlorophenyl)-1,3,4-oxadiazol-2-yl]chromone (3b)** was obtained in 71% yield; mp 202-205°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3025 (Ar–H), 1690 (C=O), 1635 (C=N), 1044 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 7.25-7.83 (8H, m, Ar–H); 6.01 (1H, s, 2-H). Mass spectrum,  $m/z$  324  $[\text{M}]^+$ . Found, %: C 62.62; H 2.83; N 8.77.  $\text{C}_{17}\text{H}_9\text{ClN}_2\text{O}_3$ . Calculated, %: C 62.88; H 2.79; N 8.63.

**3-[5-(*o*-Nitrophenyl)-1,3,4-oxadiazol-2-yl]chromone (3c)** was obtained in 68% yield; mp 222-224°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3058 (Ar–H), 1672 (C=O), 1627 (C=N), 1520, 1341 (– $\text{NO}_2$ ), 1088 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 7.34-8.12 (8H, m, Ar–H); 5.82 (1H, s, 2-H). Mass spectrum,  $m/z$  335  $[\text{M}]^+$ . Found, %: C 60.20; H 2.79; N 12.70.  $\text{C}_{17}\text{H}_9\text{N}_3\text{O}_5$ . Calculated, %: C 60.90; H 2.71; N 12.53.

**3-[5-(*p*-Nitrophenyl)-1,3,4-oxadiazol-2-yl]chromone (3d)** was obtained in 61% yield; mp 258-260°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3079 (Ar–H), 1681 (C=O), 1632 (C=N), 1498, 1319 (– $\text{NO}_2$ ), 1060 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 7.5-7.8 (8H, m, Ar–H); 5.84 (1H, s, 2-H). Mass spectrum,  $m/z$  335  $[\text{M}]^+$ . Found, %: C 61.08; H 2.74; N 12.57.  $\text{C}_{17}\text{H}_9\text{N}_3\text{O}_5$ . Calculated, %: C 60.90; H 2.71; N 12.53.

**3-[5-(2-Phenyl-1,2,3-triazol-4-yl)-1,3,4-oxadiazol-2-yl]chromone (3e)** was obtained in 50% yield; mp >280°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 2980 (Ar–H), 1697 (C=O), 1653 (C=N), 1013 (C–O–C).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 7.40-8.14 (10H, m, triazole H, Ar–H); 5.95 (1H, s, 2-H). Mass spectrum,  $m/z$  357  $[\text{M}]^+$ . Found, %: C 63.71; H 3.16; N 19.76.  $\text{C}_{19}\text{H}_{11}\text{N}_5\text{O}_3$ . Calculated, %: C 63.87; H 3.10; N 19.60.

**3-[5-(2-Trifluoromethylbenzimidazol-1-ylmethyl)-1,3,4-oxadiazol-2-yl]chromone (3f)** was obtained in 80% yield; mp 236-238°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3020 (Ar-H), 2994 (-CH<sub>2</sub>), 1688 (C=O), 1638 (C=N), 1114 (C-O-C). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.22-7.94 (8H, m, Ar-H); 5.80 (1H, s, 2-H); 5.52 (2H, s, CH<sub>2</sub>). Mass spectrum,  $m/z$  412 [M]<sup>+</sup>. Found, %: C 58.23; H 2.71; N 13.63. C<sub>20</sub>H<sub>11</sub>F<sub>3</sub>N<sub>4</sub>O<sub>3</sub>. Calculated, %: C 58.26; H 2.69; N 13.59.

**3-[5-(Benzotriazol-1-ylmethyl)-1,3,4-oxadiazol-2-yl]chromone (3g)** was obtained in 52% yield; mp 254-256°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3089 (Ar-H), 2958 (-CH<sub>2</sub>), 1694 (C=O), 1646 (C=N), 1100 (C-O-C). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.24-7.76 (8H, m, Ar-H); 5.84 (1H, s, 2-H); 5.33 (2H, s, CH<sub>2</sub>). Mass spectrum,  $m/z$  345 [M]<sup>+</sup>. Found, %: C 62.78; H 3.26; N 20.38. C<sub>18</sub>H<sub>11</sub>N<sub>5</sub>O<sub>3</sub>. Calculated, %: C 62.61; H 3.21; N 20.28.

**3-[5-(Pyridin-3-yl)-1,3,4-oxadiazol-2-yl]chromone (3h)** was obtained in 55% yield; mp 199-201°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3046 (Ar-H), 1677 (C=O), 1620 (C=N), 1081 (C-O-C). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 9.33, 8.90, 8.71, 8.12 (4H, Py-H); 7.59-7.94 (4H, m, Ar-H); 5.90 (1H, s, 2-H). Mass spectrum,  $m/z$  [M]<sup>+</sup>. Found, %: C 65.90; H 3.01; N 14.48. C<sub>16</sub>H<sub>9</sub>N<sub>3</sub>O<sub>3</sub>. Calculated, %: C 65.98; H 3.11; N 14.43.

**3-[5-(5-Methyl-1-phenyl-1,2,3-triazol-4-yl)-1,3,4-oxadiazol-2-yl]chromone (3i)** was obtained in 75% yield; mp 218-220°C. IR spectrum,  $\nu$ ,  $\text{cm}^{-1}$ : 3067 (Ar-H), 1669 (C=O), 1633 (C=N), 1056 (C-O-C). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 7.24-7.55 (9H, m, Ar-H); 6.05 (1H, s, 2-H); 2.65 (3H, s, CH<sub>3</sub>). Mass spectrum,  $m/z$  371 [M]<sup>+</sup>. Found, %: C 64.71; H 3.57; N 18.91. C<sub>20</sub>H<sub>13</sub>N<sub>5</sub>O<sub>3</sub>. Calculated, %: C 64.69; H 3.53; N 18.86.

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## REFERENCES

1. M. S. Frasinuk and V. P. Khilya, *Khim. Geterotsykl. Soedin.*, **3** (1999).
2. A. Nohara, T. Umetani, and Y. Sanno, *Tetrahedron Lett.*, **22**, 1995 (1973).
3. M. S. Gibson, *Tetrahedron*, **18**, 1377 (1962).
4. L. M. Chen, Z. Y. Zhang, X. Zhang, et al., *Chem. J. Chin. Univ.*, **9**, 283 (1988).